

Subject:- Science

Class- VIII

Chapter-4

MATERIALS : METALS AND NON-METALS

Exercises:

1. **Which of the following can be beaten into thin sheets?**

- (a) Zinc (b) Phosphorus (c) Sulphur (d) Oxygen

Ans: (a) Zinc

2. **Which of the following statements is correct?**

- (a) All metals are ductile.
(b) All non-metals are ductile.
(c) Generally, metals are ductile.
(d) Some non-metals are ductile.

Ans:- (c) **Generally, metals are ductile.**

3. **Fill in the blanks.**

- (a) Phosphorus is a very ----- non-metal.
(b) Metals are ----- conductors of heat and ----- .
(c) Iron is ----- reactive than copper.
(d) Metals react with acids to produce ----- gas.

Ans: (a) **reactive**, (b) **good & electricity**, (c) **more**
(d) **hydrogen.**

4. **Mark 'T' if the statement is true and 'F' if it is false.**

- (a) Generally, non-metals react with acids. ()
(b) Sodium is a very reactive metal. ()
(c) Copper displaces zinc from zinc sulphate solution. ()
(d) Coal can be drawn into wires. ()

Ans: (a) \rightarrow (x), (b) \rightarrow (v), (c) \rightarrow (x), (d) \rightarrow (x)

5. Some properties are listed in the following Table. Distinguish between metals and non-metals on the basis of these properties.

Properties	Metals	Non-metals
1. Appearance	Lustrous	Not-Lustrous
2. Hardness	Hard	Soft
3. Malleability	Highly Malleable	Not Malleable
4. Ductility	Most ductile	Not ductile (These are brittle)
5. Heat Conduction	Good conductor of heat	Poor conductor of heat
6. Conduction of Electricity	Good conductor of electricity	The bad conductor of electricity

6. Give reasons for the following.

(a) Aluminium foils are used to wrap food items.

Ans:- **Aluminium is malleable and can be drawn into thin sheets hence Aluminium foils are used to wrap food items.**

(b) Immersion rods for heating liquids are made up of metallic substances.

Ans: Immersion rods for heating liquids are made up of metallic substances because metals are good conductors of heat and electricity.

(c) Copper cannot displace zinc from its salt solution.

Ans: Copper cannot displace zinc from its salt solution because Zinc is more reactive than copper.

(d) Sodium and potassium are stored in kerosene.

Ans: Sodium and Potassium are highly reactive metals which readily reacts with atmospheric Oxygen to catch fire hence Sodium and Potassium are stored in kerosene.

7. Can you store lemon pickle in an aluminium utensil? Explain.

Ans: Pickle consists of acids which react with Aluminium metal to produce salt and Hydrogen. Hence pickle is not stored in aluminium utensil.

8. Match the substances given in Column 'A' with their uses given in Column 'B'.

A	B
(i) Gold	(a) Thermometers
ii) Iron	(b) Electric wire
(iii) Aluminium	(c) Wrapping food
iv) Carbon	(d) Jewellery
(v) Copper	(e) Machinery
(vi) Mercury	(f) Fuel

Ans:- (i) → (d), (ii) → (e), (iii) → (c), (iv) → (f),
(v) → (b), (vi) → (a).

9. What happens when:

(a) Dilute sulphuric acid is poured on a copper plate?

Ans: No reaction occurs when dilute sulphuric acid is poured on a copper plate. However, when concentrated sulphuric acid is poured on a copper plate, hydrogen gas evolves along with the formation of blue coloured copper sulphate crystals. The chemical reaction for the reaction between concentrated sulfuric acid and copper is:



(b) Iron nails are placed in copper sulphate solution?

Write word equations of the reactions involved.

Ans: Iron being more reactive displaces copper from copper sulphate. In this reaction, the blue colour of copper sulphate fades and there is a deposition of copper on the iron nail. $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$

10. Saloni took a piece of burning charcoal and collected the gas evolved in a test tube.

(a) How will she find the nature of the gas ?

Ans: In a test tube containing gas, add a few drops of water. Now cover the test tube and shake well. After shaking, test the solution with blue litmus. It will change from blue to red. Thus, gas is acidic in nature.

(b) Write down word equations of all the reactions taking place in this process.

Ans: Charcoal reacts with oxygen to form carbon dioxide gas.



(Carbon from charcoal)

(Oxygen)

(Carbon dioxide)

11. One day Reeta went to a jeweller's shop with her an old gold jewellery to the goldsmith to polish. Next day when they brought the jewellery back, they found that there was a slight loss in its weight. Can you suggest a reason for the loss in weight?

Ans:- In order to polish the gold ornament, it is to be dipped into a liquid called aqua regia (a mixture of hydrochloric acid and nitric acid). On getting dissolved in the environment of aqua regia, the outer layer of gold dissolves and an inner shiny layer appears. The dissolving of the layer causes a reduction in the weight of the jewellery.

-----Thanks-----

